Nano-confined Amorphous Solid-oxidant Cathodes for Next-generation Lithium Ion Battery
Technology #17959

Applications
Nano-confined amorphous solid-oxidant cathodes improve lithium battery performance primarily by increasing the number of lithium ion sites on the cathode. Lithium batteries have applications ranging from portable electronics to electric vehicles.

Problem Addressed
Traditional crystalline oxide cathodes, such as LiCoO2+, LiMn2O4, and LiFePO4, etc., have low practical capacity (<200 mAhg-1) and cannot satisfy the demands of the increasing capacity of lithium ion batteries. Li2S nanoparticles synthesized and dispersed on reduced graphene oxide cathode paper for a high-performance lithium sulfur battery achieve a high reversible capacity of 597mAhg-1 at a high rate of 7C, which proves that Nanosized Amorphous Solid-Oxidant Cathodes (NASOCs) can improve rate capability and lower over-potential.

Technology
These anodes can be composed of several different materials including unencapsulated amorphous Ca(ClO)2 or nanolithia (Li2O/LiO2) encapsulated by Co3O4. The advantage of these amorphous structures is that they have more sites to accommodate Li ions than traditional crystalline electrodes. The number of Li ion sites is directly related to the electrode's capacity; therefore, the increased number of Li ion sites increases the battery's overall capacity. NASOCs are also self-contained without need to exchange oxygen with the outside (only lithium goes in and out), which makes them more stable and reliable. These cathodes also have a lower overpotential (0.25V compared to 1V) - potential difference between charging and discharging - which reduces heat emitted by the battery.

Advantages
- Improves reliability and charging time
- Increases battery voltage and capacity

Categories For This Invention:
Energy
Energy Storage
Batteries
Lithium Batteries
Materials
Micro & Nanotech
Nanomaterials

Intellectual Property:

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US Patent Pending

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Publications:

Anion-redox Nanolithia Cathodes for Li-ion Batteries
Nature Energy
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External Links:

Li Group
http://li.mit.edu/

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